

JEE (Main)-2026 Session-1
Question Paper with Solutions
(Mathematics, Physics, And Chemistry)
28 January 2026 Shift – 1

Time: 3 hrs.

M.M: 300

IMPORTANT INSTRUCTIONS:

- (1) The test is of 3 hours duration.
- (2) This test paper consists of 75 questions. Each subject (PCM) has 25 questions. The maximum marks are 300.
- (3) This question paper contains Three Parts. Part-A is Physics, Part-B is Chemistry and Part-C is Mathematics. Each part has only two sections: Section-A and Section-B.
- (4) Section - A: Attempt all questions.
- (5) Section - B: Attempt all questions.
- (6) Section - A (01 - 20) contains 20 multiple choice questions which have only one correct answer. Each question carries +4 marks for correct answer and -1 mark for wrong answer.
- (7) Section - B (21 - 25) contains 5 Numerical value-based questions. The answer to each question should be rounded off to the nearest integer. Each question carries +4 marks for correct answer and -1 mark for wrong answer.

4. If $\frac{\tan(A-B)}{\tan A} + \frac{\sin^2 C}{\sin^2 A} = 1$, $A, B, C \in \left(0, \frac{\pi}{2}\right)$,

then

- (1) $\tan A, \tan C, \tan B$ are in G.P.
- (2) $\tan A, \tan B, \tan C$ are in G.P.
- (3) $\tan A, \tan C, \tan B$ are in A.P.
- (4) $\tan A, \tan B, \tan C$ are in A.P.

Ans. (1)

Sol. $\frac{\tan A - \tan B}{(1 + \tan A \tan B) \tan A} + \frac{1 + \cot^2 A}{1 + \cot^2 C} = 1$

Put $\tan A = x, \tan B = y, \tan C = z$

$$\therefore \frac{x - y}{(1 + xy)x} + \frac{(x^2 + 1)z^2}{x^2(z^2 + 1)} = 1$$

$$\therefore x(x - y)(z^2 + 1) + z^2(1 + x^2)(1 + xy) = (1 + xy)x^2(1 + z^2)$$

after solving we get

$$z^2 = xy \quad \because 1 + x^2 \neq 0$$

$$\therefore \tan^2 C = \tan A \cdot \tan B$$

$\tan A, \tan C, \tan B$ are in G.P.

5. Let z be a complex number such that $|z - 6| = 5$ and $|z + 2 - 6i| = 5$.

Then the value of $z^3 + 3z^2 - 15z + 141$ is equal to

- (1) 42
- (2) 37
- (3) 50
- (4) 61

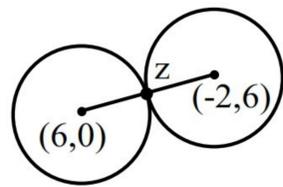
Ans. (3)

Sol. Center of first circle $C_1(6,0), r_1 = 5$

Center of second circle $C_2(-2,6), r_2 = 5$

$$\because C_1 C_2 = r_1 + r_2$$

\therefore common point Z is mid point of C_1 & C_2



$$\therefore z = 2 + 3i$$

$$\therefore z^2 = 4z - 13$$

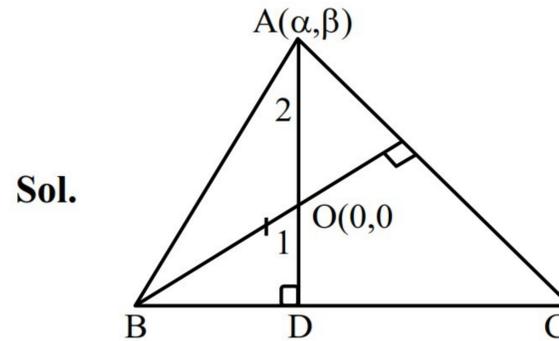
$$\therefore z^3 = 3z - 52$$

$$\therefore z^3 + 3z^2 - 15z + 141 = 50$$

6. Let ABC be an equilateral triangle with orthocenter at the origin and the side BC on the line $x + 2\sqrt{2}y = 4$. If the co-ordinates of the vertex A are (α, β) , then the greatest integer less than or equal to $|\alpha + \sqrt{2}\beta|$ is

- (1) 2
- (2) 3
- (3) 5
- (4) 4

Ans. (4)



Sol.

$$\because m_{BC} \cdot m_{AD} = -1$$

$$\Rightarrow \left(-\frac{1}{2\sqrt{2}}\right) \left(\frac{\beta}{\alpha}\right) = -1$$

$$\Rightarrow \beta = 2\sqrt{2}\alpha \quad \dots(1)$$

$$\because OD = \left|\frac{-4}{\sqrt{1+8}}\right| = \frac{4}{3} \Rightarrow AO = \frac{8}{3}$$

$$\text{So } AD = \frac{8}{3} + \frac{4}{3} = 4$$

$$\Rightarrow \frac{|\alpha + 2\sqrt{2}\beta - 4|}{3} = 4 \Rightarrow \alpha = \frac{16}{9} \text{ or } -\frac{8}{9}$$

$\left\{ \because A(\alpha, \beta) \text{ \& } (0,0) \text{ lies on same side of given line} \right\}$

$$\therefore (\alpha, \beta) = \left(\frac{16}{9}, \frac{32\sqrt{2}}{9}\right); \text{ (Rejected)}$$

$$\text{so } (\alpha, \beta) = \left(-\frac{8}{9}, -\frac{16\sqrt{2}}{9}\right)$$

$$= \left[|\alpha + \sqrt{2}\beta|\right] = \left[\left|-\frac{8}{9} - \frac{32}{9}\right|\right] = 4$$

7. Let $S = \{1, 2, 3, 4, 5, 6, 7, 8, 9\}$. Let x be the number of 9-digit numbers formed using the digits of the set S such that only one digit is repeated and it is repeated exactly twice. Let y be the number of 9-digit numbers formed using the digits of the set S such that only two digits are repeated and each of these is repeated exactly twice. Then,

- (1) $29x = 5y$ (2) $45x = 7y$
 (3) $21x = 4y$ (4) $56x = 9y$

Ans. (3)

Sol. $S = \{1, 2, 3, \dots, 9\}$

$$x = {}^9C_1 \cdot {}^8C_7 \times \frac{9!}{2} = \frac{9 \times 8 \times 9!}{2}$$

$$y = {}^9C_2 \cdot {}^7C_5 \times \frac{9!}{2! \times 2!} = \frac{9 \times 8}{2} \times \frac{7 \times 6}{2} \times \frac{9!}{2! \times 2!}$$

$$\Rightarrow \frac{x}{y} = \frac{4}{21}$$

$$21x = 4y$$

8. Let $S = \{x^3 + ax^2 + bx + c : a, b, c \in \mathbb{N} \text{ and } a, b, c \leq 20\}$ be a set of polynomials. Then the number of polynomials in S , which are divisible by $x^2 + 2$, is

- (1) 20 (2) 6
 (3) 120 (4) 10

Ans. (4)

Sol. $x^3 + ax^2 + bx + c = (x^2 + 2)\left(x + \frac{c}{2}\right)$

$$x^2 : a = \frac{c}{2}$$

$$x : b = 2$$

$$b = 2, a = \frac{c}{2} \quad c \in \{2, 4, \dots, 20\}$$

Number of polynomials in 'S' will be 10.

9. A bag contains 10 balls out of which k are red and $(10 - k)$ are black, where $0 \leq k \leq 10$. If three balls are drawn at random without replacement and all of them are found to be black, then the probability that the bag contains 1 red and 9 black balls is :

- (1) $\frac{7}{11}$ (2) $\frac{7}{55}$
 (3) $\frac{7}{110}$ (4) $\frac{14}{55}$

Ans. (4)

Sol. Probability = $\frac{{}^1C_0 \cdot {}^9C_3}{\sum_{k=0}^{10} {}^kC_0 \cdot {}^{10-k}C_3}$

$$= \frac{{}^9C_3}{{}^{10}C_3 + {}^9C_3 + {}^8C_3 + \dots + {}^3C_3}$$

$$= \frac{{}^9C_3}{{}^{11}C_4} = \frac{14}{55}$$

10. **Ans. ()**

Sol. $\frac{\beta - \alpha}{\alpha\beta} = \frac{1}{3}, \quad \alpha + \beta = \frac{\lambda + 3}{\lambda}, \quad \alpha\beta = \frac{3}{\lambda}$

$$\beta - \alpha = \frac{\alpha\beta}{3} = \frac{1}{\lambda}$$

on squaring

$$\alpha^2 + \beta^2 - 2\alpha\beta = \frac{1}{\lambda^2} \quad \dots\dots\dots(1)$$

$$\alpha^2 + \beta^2 + 2\alpha\beta = \frac{(\lambda + 3)^2}{\lambda^2} \quad \dots\dots\dots(2)$$

$$(2) - (1) \quad 4\alpha\beta = \frac{(\lambda + 3)^2 - 1}{\lambda^2}$$

$$\frac{12}{\lambda} = \frac{\lambda^2 + 6\lambda + 8}{\lambda^2}$$

$$\Rightarrow \lambda^2 - 6\lambda^2 + 8\lambda = 0$$

$$\Rightarrow \lambda = 0, 2, 4$$

Sum of possible values of λ is = 6

10. If α, β , where $\alpha < \beta$, are the roots of the equation $\lambda x^2 - (\lambda + 3)x + 3 = 0$ such that $\frac{1}{\alpha} - \frac{1}{\beta} = \frac{1}{3}$, then

the sum of all possible values of λ is :

- (1) 6 (2) 2
 (3) 4 (4) 8

Ans. (1)

Sol. $\frac{\beta - \alpha}{\alpha\beta} = \frac{1}{3}, \alpha + \beta = \frac{\lambda + 3}{\lambda}, \alpha\beta = \frac{3}{\lambda}$

$$\beta - \alpha = \frac{\alpha\beta}{3} = \frac{1}{\lambda}$$

on squaring

$$\alpha^2 + \beta^2 - 2\alpha\beta = \frac{1}{\lambda^2} \dots\dots\dots(1)$$

$$\alpha^2 + \beta^2 + 2\alpha\beta = \frac{(\lambda + 3)^2}{\lambda^2} \dots\dots\dots(2)$$

$$(2) - (1) \quad 4\alpha\beta = \frac{(\lambda + 3)^2 - 1}{\lambda^2}$$

$$\frac{12}{\lambda} = \frac{\lambda^2 + 6\lambda + 8}{\lambda^2}$$

$$\Rightarrow \lambda^2 - 6\lambda^2 + 8\lambda = 0$$

$$\Rightarrow \lambda = 0, 2, 4$$

Sum of possible values of λ is = 6

11. If $\int \left(\frac{1 - 5\cos^2 x}{\sin^5 x \cos^2 x} \right) dx = f(x) + C$ where C is the

constant of integration, then $f\left(\frac{\pi}{6}\right) - f\left(\frac{\pi}{4}\right)$ is equal

to

(1) $\frac{1}{\sqrt{3}}(26 + \sqrt{3})$ (2) $\frac{4}{\sqrt{3}}(8 - \sqrt{6})$

(3) $\frac{1}{\sqrt{3}}(26 - \sqrt{3})$ (4) $\frac{2}{\sqrt{3}}(4 + \sqrt{6})$

Ans. (2)

Sol. $\int \frac{dx}{\sin^5 x \cos^2 x} - 5 \int \frac{dx}{\sin^5 x}$
 $\int \frac{\sec^2 x dx}{\sin^5 x} - 5 \int \frac{dx}{\sin^5 x}$

By IBP

$$= \frac{\tan x}{\sin^5 x} - \int -\frac{5}{\sin^6 x} \cdot \cos x \cdot \tan x dx - 5 \int \frac{dx}{\sin^5 x}$$

$$= \frac{\tan x}{\sin^5 x} + c$$

$$f(x) = \frac{\tan x}{\sin^5 x}$$

$$f\left(\frac{\pi}{6}\right) - f\left(\frac{\pi}{4}\right) = \frac{2^5}{\sqrt{3}} - (\sqrt{2})^5 = 4\sqrt{2} - \frac{32}{\sqrt{3}}$$

$$= \frac{32}{\sqrt{3}} - 4\sqrt{2}$$

$$= \frac{4}{\sqrt{3}}(8 - \sqrt{6})$$

12. Let f be a polynomial function such that $f(x^2 + 1) = x^4 + 5x^2 + 2$, for all $x \in \mathbb{R}$.

Then $\int_0^3 f(x) dx$ is equal to

(1) $\frac{41}{3}$ (2) $\frac{33}{2}$

(3) $\frac{27}{2}$ (4) $\frac{5}{3}$

Ans. (2)

Sol. $\because f(x^2 + 1) = x^4 + 5x^2 + 2$
 {put $x^2 + 1 = t$ }

$$\Rightarrow f(t) = (t - 1)^2 + 5(t - 1) + 2$$

$$\Rightarrow f(t) = t^2 + 3t - 2$$

$$\text{Now, } \int_0^3 f(t) dt = \int_0^3 (t^2 + 3t - 2) dt$$

$$\left[\frac{t^3}{3} + \frac{3t^2}{2} - 2t \right]_0^3$$

$$\left[\frac{27}{3} + \frac{27}{2} - 6 \right]$$

$$= \frac{33}{2}$$

16. Let A, B and C be three 2×2 matrices with real entries such that $B = (I + A)^{-1}$ and $A + C = I$. If $BC = \begin{bmatrix} 1 & -5 \\ -1 & 2 \end{bmatrix}$ and $CB \begin{bmatrix} x_1 \\ x_2 \end{bmatrix} = \begin{bmatrix} 12 \\ -6 \end{bmatrix}$, then

$x_1 + x_2$ is

- (1) 2 (2) 0
(3) -2 (4) 4

Ans. (2)

Sol. $B = (I + A)^{-1}$, $A + C = I$

$$\Rightarrow B(I + A) = (I + A)B = I$$

$$\Rightarrow B + BA = B + AB$$

$$\Rightarrow B + B(I - C) = B + (I - C)B$$

$$\Rightarrow 2B - BC = 2B - CB$$

$$\Rightarrow BC = CB$$

$$\therefore CB \begin{bmatrix} x_1 \\ x_2 \end{bmatrix} = \begin{bmatrix} 1 & -5 \\ -1 & 2 \end{bmatrix} \begin{bmatrix} x_1 \\ x_2 \end{bmatrix} = \begin{bmatrix} 12 \\ -6 \end{bmatrix}$$

$$\Rightarrow \begin{bmatrix} x_1 \\ x_2 \end{bmatrix} = \begin{bmatrix} 1 & -5 \\ -1 & 2 \end{bmatrix}^{-1} \begin{bmatrix} 12 \\ -6 \end{bmatrix} = -\frac{1}{3} \begin{bmatrix} 2 & 5 \\ 1 & 1 \end{bmatrix} \begin{bmatrix} 32 \\ -6 \end{bmatrix}$$

$$\Rightarrow \begin{bmatrix} x_1 \\ x_2 \end{bmatrix} = \begin{bmatrix} 2 \\ -2 \end{bmatrix} \therefore x_1 + x_2 = 0$$

17. The common difference of the A.P.: a_1, a_2, \dots, a_m is 13 more than the common difference of the A.P.: b_1, b_2, \dots, b_n . If $b_{31} = -277$, $b_{43} = -385$ and $a_{78} = 327$, then a_1 is equal to

- (1) 21 (2) 24
(3) 19 (4) 16

Ans. (3)

Sol. Let common difference of A.P.'s are d_1 & d_2

$$\therefore d_1 = 13 + d_2$$

$$b_1 + 30d_2 = -277 \quad \dots(1)$$

$$b_1 + 42d_2 = -385 \quad \dots(2)$$

By (2) - (1)

$$12d_2 = -108$$

$$d_2 = -9$$

$$\therefore \boxed{d_1 = 4}$$

Now $a_{78} = 327$

$$\Rightarrow a_1 + 77d_1 = 327$$

$$\Rightarrow a_1 + 308 = 327$$

$$a_1 = 19$$

18. If the distances of the point $(1, 2, a)$ from the line $\frac{x-1}{1} = \frac{y}{2} = \frac{z-1}{1}$ along the lines

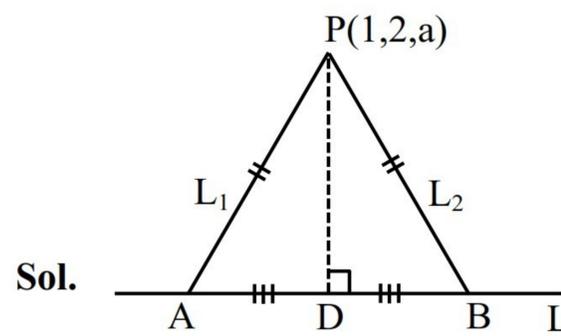
$$L_1: \frac{x-1}{3} = \frac{y-2}{4} = \frac{z-a}{b} \text{ and}$$

$$L_2: \frac{x-1}{1} = \frac{y-2}{4} = \frac{z-a}{c} \text{ are equal,}$$

then $a + b + c$ is equal to

- (1) 7 (2) 5
(3) 6 (4) 4

Ans. (1)



Sol.

$$L: \frac{x-1}{1} = \frac{y}{2} = \frac{z-1}{1}$$

$$L_1: \frac{x-1}{3} = \frac{y-2}{4} = \frac{z-a}{b} = \lambda$$

$$L_2: \frac{x-1}{1} = \frac{y-2}{4} = \frac{z-a}{c} = \mu$$

Let $A(3\lambda + 1, 4\lambda + 2, b\lambda + a)$

It lies on L

$$\therefore \frac{3\lambda}{1} = \frac{4\lambda + 2}{2} = \frac{b\lambda + a - 1}{1}$$

$$\Rightarrow \boxed{\lambda = 1} \text{ and } a + b - 1 = 3$$

$$\Rightarrow A(4, 6, 4), \quad a + b = 4 \quad \dots(1)$$

Let $B(\mu + 1, 4\mu + 2, c\mu + a)$

It also lies on L

$$\frac{\mu}{1} = \frac{4\mu + 2}{2} = \frac{c\mu + a - 1}{1}$$

$$\Rightarrow 2\mu = 4\mu + 2$$

$$\Rightarrow \boxed{\mu = -1}$$

$$a - c - 1 = -1$$

$$\Rightarrow \boxed{a = c} \quad \dots(2) \quad \& \quad B(0, -2, 0)$$

22. For some $\theta \in \left(0, \frac{\pi}{2}\right)$, let the eccentricity and the length of the latus rectum of the hyperbola $x^2 - y^2 \sec^2 \theta = 8$ be e_1 and l_1 , respectively, and let the eccentricity and the length of the latus rectum of the ellipse $x^2 \sec^2 \theta + y^2 = 6$ be e_2 and l_2 , respectively. If $e_1^2 = e_2^2 (\sec^2 \theta + 1)$, then $\left(\frac{l_1 l_2}{e_1 e_2}\right) \tan^2 \theta$ is equal to _____.

Ans. (8)

Sol. $\frac{x^2}{8} - \frac{y^2}{8 \cos^2 \theta} = 1$, $e_1 = \sqrt{1 + \frac{8 \cos^2 \theta}{8}}$
 $l_1 = \frac{2b^2}{a} = \frac{2 \cdot (8 \cos^2 \theta)}{2\sqrt{2}}$
 $\frac{x^2}{6} + \frac{y^2}{6 \cos^2 \theta} = 1$; $e_2 = \sqrt{1 - \frac{6 \cos^2 \theta}{6}} = \sin \theta$
 $l_2 = \frac{2b^2}{a} = \frac{2 \cdot 6 \cos^2 \theta}{\sqrt{6}}$
 $e_1^2 = e_2^2 (1 + \sec^2 \theta)$
 $1 + \cos^2 \theta = \sin^2 \theta \left(1 + \frac{1}{\cos^2 \theta}\right)$
 $1 + \cos^2 \theta = \sin^2 \theta + \tan^2 \theta$
 Solving we get $\theta = \frac{\pi}{4}$
 $l_1 = 2\sqrt{2}$
 $e_1 = \sqrt{\frac{3}{2}}$
 $l_2 = \sqrt{6}$
 $e_2 = \frac{1}{\sqrt{2}}$
 $\left(\frac{l_1 l_2}{e_1 e_2}\right) \tan^2 \theta = 8$ (By putting values)

23. If $k = \tan\left(\frac{\pi}{4} + \frac{1}{2} \cos^{-1}\left(\frac{2}{3}\right)\right) + \tan\left(\frac{1}{2} \sin^{-1}\left(\frac{2}{3}\right)\right)$ then the number of solutions of the equation $\sin^{-1}(kx - 1) = \sin^{-1} x - \cos^{-1} x$ is _____.

Ans. (1)

Sol. Let $\theta = \frac{1}{2} \sin^{-1} \frac{2}{3}$, then $\frac{1}{2} \cos^{-1} \frac{1}{3} = \left(\frac{\pi}{4} - \theta\right)$

$$k = \tan \theta + \cot \theta = \frac{1}{\sin \theta \cos \theta} = \frac{2}{\sin 2\theta}$$

$$k = \frac{2}{\frac{2}{3}} = 3$$

$$\sin^{-1}(3x - 1) = \sin^{-1} x - \cos^{-1} x$$

$$\sin^{-1}(3x - 1) = \frac{\pi}{2} - 2\cos^{-1} x$$

$$3x - 1 = \sin\left(\frac{\pi}{2} - 2\cos^{-1} x\right)$$

$$3x - 1 = 2x^2 - 1 \Rightarrow x = 0, \frac{3}{2} \text{ (rejected)}$$

No. of solution = 1

24. The value of $\sum_{r=1}^{20} \left(\left| \sqrt{\pi \left(\int_0^r x |\sin \pi x| dx \right)} \right| \right)$ is _____.

Ans. (210)

Sol. Let $I_r = \int_0^r x |\sin \pi x| dx$ (1)

Apply King Property

$$= \int_0^r (r - x) |\sin \pi x| dx \quad \dots(2)$$

By (1) + (2)

$$2I_r = \int_0^r r |\sin \pi x| dx \Rightarrow I_r = \frac{r}{2} \int_0^r |\sin \pi x| dx$$

$$I_1 = \frac{1}{2} \int_0^1 |\sin \pi x| dx = \frac{1}{2\pi} \int_0^\pi |\sin t| dt = \frac{1}{2\pi} (2)$$

$$I_2 = \frac{2}{2} \int_0^2 |\sin \pi x| dx = \frac{2}{2\pi} \int_0^{2\pi} |\sin t| dt = \frac{2}{2\pi} (4)$$

$$S = \sqrt{\pi \cdot \frac{1}{2\pi} \cdot 2} + \sqrt{\pi \cdot \frac{2}{2\pi} \cdot 4} + \sqrt{\pi \cdot \frac{3}{2\pi} \cdot 6} + \dots + \sqrt{\pi \cdot \frac{20}{2\pi}} \quad (2.20)$$

$$= 1 + 2 + 3 + \dots + 20$$

$$= \frac{20 \times 21}{2} = 210$$

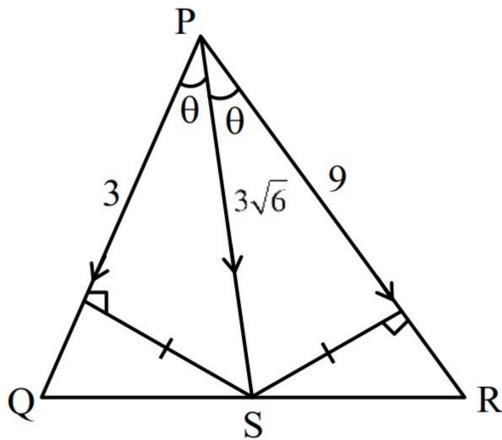
25. Let PQR be a triangle such that $\vec{PQ} = -2\hat{i} - \hat{j} + 2\hat{k}$ and $\vec{PR} = a\hat{i} + b\hat{j} - 4\hat{k}$, $a, b \in \mathbb{Z}$. Let S be the point on QR, which is equidistant from the lines PQ and PR. If $|\vec{PR}| = 9$ and $\vec{PS} = \hat{i} - 7\hat{j} + 2\hat{k}$, then the value of $3a - 4b$ is ____.

Ans. (37)

Sol. $\vec{PQ} = -2\hat{i} - \hat{j} + 2\hat{k}$

$$\vec{PR} = a\hat{i} + b\hat{j} - 4\hat{k} \quad (a, b \in \mathbb{Z})$$

$$\vec{PS} = \hat{i} - 7\hat{j} + 2\hat{k}$$



$$|\vec{PR}| = 9$$

$$a^2 + b^2 + 16 = 81$$

$$a^2 + b^2 = 65 \quad \dots(1)$$

$$\cos \theta = \frac{\vec{PQ} \cdot \vec{PS}}{|\vec{PQ}| |\vec{PS}|}$$

$$= \frac{-2 + 7 + 4}{3 \cdot 3\sqrt{6}} = \frac{9}{3 \cdot 3\sqrt{6}}$$

$$= \frac{1}{\sqrt{6}}$$

$$\frac{1}{\sqrt{6}} = \frac{\vec{PS} \cdot \vec{PR}}{|\vec{PS}| |\vec{PR}|} = \frac{a - 7b - 8}{3\sqrt{6} \cdot 9}$$

$$a - 7b = 35 \quad \dots(2)$$

From (1) & (2)

$$\Rightarrow a = 7, b = -4$$

$$\therefore 3a - 4b = 21 + 16 = 37$$

(Matched with NTA)

But here PS is internal Angle bisector of $\angle QPR$ hence

$$PS = \frac{2PQ \cdot PR}{PQ + PR} \cos \theta$$

$$\Rightarrow 3\sqrt{6} = \frac{2 \times 3 \times 9}{3 + 9} \cos \theta$$

$$\Rightarrow \cos \theta = \frac{2\sqrt{6}}{3} = \frac{2\sqrt{2}}{\sqrt{3}} > 1 \quad (\text{Not possible})$$

Hence NO such triangle is possible

Another way to check

For evaluated value of a & b [$a = 7, b = -4$]

$$\vec{PR} = 7\hat{i} - 4\hat{j} - 4\hat{k}$$

$$\vec{PQ} = -2\hat{i} - \hat{j} + 2\hat{k}$$

$$\vec{PS} = \hat{i} - 7\hat{j} + 2\hat{k}$$

$$\therefore \vec{QS} = \vec{PS} - \vec{PQ} = 3\hat{i} - 6\hat{j}$$

$$\vec{SR} = \vec{PR} - \vec{PS} = 6\hat{i} + 3\hat{j} - 6\hat{k}$$

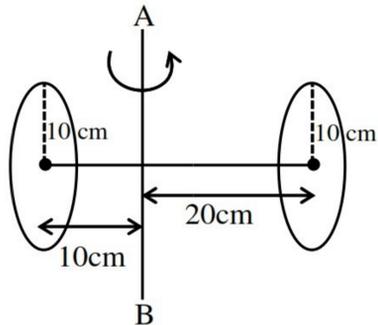
$$\therefore \vec{QS} \text{ is not parallel to } \vec{SR}$$

$$\therefore Q, S, R \text{ are not co-linear}$$

(Bonus)

29. Two circular discs of radius each 10 cm are joined at their centres by a rod of length 30 cm and mass 600 gm as shown in figure.

If the mass of each disc is 600 gm and applied torque between two discs is 43×10^5 dyne cm, the angular acceleration of the discs about the given axis AB is _____ rad/s^2 .



- (1) 22 (2) 11
(3) 100 (4) 27

Ans. (2)

Sol. $\alpha = \frac{\tau}{I}$

$$I = \frac{1}{4}mR^2 + mR^2 + \frac{1}{4}mR^2 + m(2R)^2 + \frac{m(3R)^2}{12} + m\left(\frac{R}{2}\right)^2$$

$$= \left(\frac{3}{2} + 4 + 1\right)mR^2 = \frac{13}{2}mR^2 = \frac{13}{2} \times 600 \times 10^2 = 39 \times 10^4$$

$$\alpha = \frac{43 \times 10^5}{39 \times 10^4} \text{ rad/s}^2 = \frac{430}{39} \text{ rad/s}^2 \approx 11 \text{ rad/s}^2$$

30. Water drops fall from a tap on the floor, 5 m below, at regular intervals of time, the first drop strikes the floor when the sixth drop begins to fall. The height at which the fourth drop will be from ground, at the instant when the first drop strikes the ground is _____ m.

$$(g = 10 \text{ m/s}^2)$$

- (1) 2.5 (2) 4.0
(3) 4.2 (4) 3.8

Ans. (3)

Sol. Time to reach ground $= \sqrt{\frac{2h}{g}} = \sqrt{\frac{2 \times 5}{10}} = 1 \text{ sec}$

Five drops per second

Time between each drop = 0.2 sec.

Time of fall for 4th drop is $1 - 0.6 = 0.4 \text{ sec}$

$$\text{Height fall of 4}^{\text{th}} \text{ drop is } = \frac{1}{2} \times 10 \times 0.4^2 = 0.8 \text{ m}$$

$$\text{Height from ground} = 5 - 0.8 = 4.2 \text{ m}$$

31. An atom ${}_3X$ is bombarded by shower of fundamental particles and in 10 s this atom absorbed 10 electrons, 10 protons and 9 neutrons. The percentage growth in the surface area of the nucleons is recorded by :

- (1) 250% (2) 150%
(3) 225% (4) 900%

NTA Ans. (3)

Allen Ans. (BONUS)

Sol. Surface area $x \propto A^{2/3}$

$$X_i = 8^{2/3} K = 4K$$

$$X_f = (8 + 10 + 9)^{2/3} K = 9K$$

% increase in surface area of nucleus

$$x_i = \frac{9K - 4K}{4K} \times 100 = 125\%$$

32. The electric field of an electromagnetic wave travelling through a medium is given by

$$\vec{E}(x, t) = 25 \sin(2.0 \times 10^{15} t - 10^7 x) \hat{n}$$

then the refractive index of the medium is _____ .

(All given measurement are in SI units)

- (1) 1.2 (2) 2
(3) 1.5 (4) 1.7

Ans. (3)

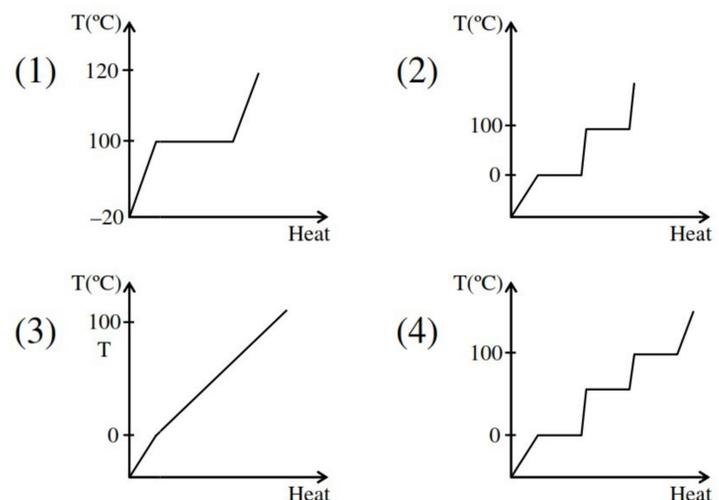
Sol. $\omega = 2 \times 10^{15} \text{ rad/s}$

$$k = 10^7 \text{ m}^{-1}$$

$$v = \frac{2\pi}{k} \cdot \frac{\omega}{2\pi} = \frac{\omega}{k} = \frac{2 \times 10^{15}}{10^7} = 2 \times 10^8 = \frac{c}{1.5}$$

$$\Rightarrow \mu = 1.5$$

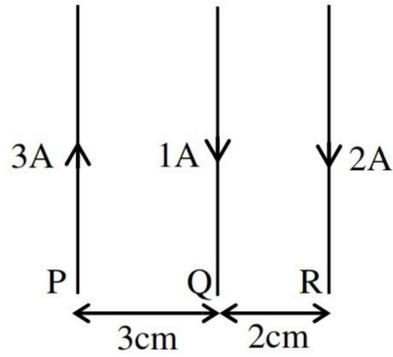
33. Which of the following best represents the temperature versus heat supplied graph for water, in the range of -20°C to 120°C ?



Ans. (2)

Sol. 2

34. Three long straight wires carrying current are arranged mutually parallel as shown in the figure. The force experienced by 15 cm length of wire Q is _____ .



$$(\mu_0 = 4\pi \times 10^{-7} \text{ T.m/A})$$

- (1) 6×10^{-7} N towards P
 (2) 6×10^{-6} N towards R
 (3) 6×10^{-7} N towards R
 (4) 6×10^{-6} N towards P

Ans. (2)

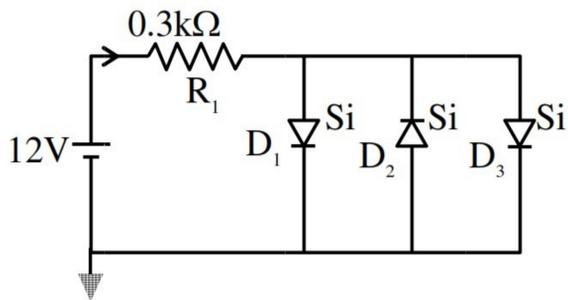
$$\text{Sol. } F_{\text{net}} = \frac{\mu_0}{2\pi} I_0 \left(\frac{I_1}{d_1} + \frac{I_2}{d_2} \right) \ell$$

$$F_{\text{net}} = 2 \times 10^{-7} \times 1 \left(\frac{3}{3} + \frac{2}{2} \right) \times \frac{15 \times 10^{-2}}{10^{-2}}$$

$$= 4 \times 15 \times 10^{-7}$$

$$F_{\text{net}} = 6 \times 10^{-6} \text{ N}$$

35. Assuming in forward bias condition there is a voltage drop of 0.7 V across a silicon diode, the current through diode D_1 in the circuit is _____ mA. (Assume all diodes in the given circuit are identical)



- (1) 20.15 (2) 11.7
 (3) 17.6 (4) 18.8

Ans. (4)

$$\text{Sol. } 12 - 0.3 \times 10^3 I - 0.7 = 0$$

$$\frac{11.3}{0.3 \times 10^3} = I$$

$$37.66 \times 10^{-3} \text{ A} = I$$

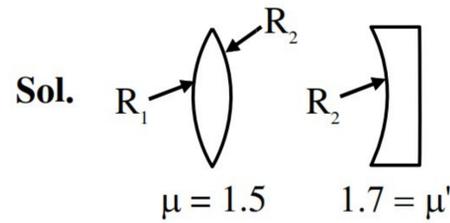
Current through diode D_1 , $I_1 = I/2$

$$I_1 = 18.83 \text{ mA}$$

36. The magnitudes of power of a biconvex lens (refractive index 1.5) and that of a plano-concave lens (refractive index = 1.7) are same. If the curvature of plano-concave lens exactly matches with the curvature of back surface of the biconvex lens, then ratio of radius of curvature of front and back surface of the biconvex lens is _____ .

- (1) 5 : 2 (2) 5 : 12
 (3) 12 : 5 (4) 2 : 5

Ans. (1)



Sol.

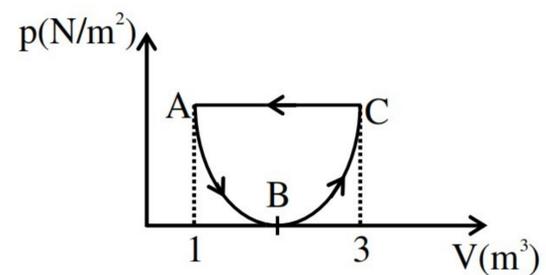
$$|P_A| = |P_B|$$

$$0.5 \left(\frac{1}{R_1} + \frac{1}{R_2} \right) = \frac{0.7}{R_2}$$

$$\frac{5}{R_1} = \frac{2}{R_2}$$

$$\frac{R_1}{R_2} = \frac{5}{2}$$

37. In the following p - V diagram the equation of state along the curved path is given by $(V - 2)^2 = 4ap$ where a is a constant. The total work done in the closed path is



- (1) $-\frac{1}{a}$ (2) $+\frac{1}{3a}$
 (3) $\frac{1}{2a}$ (4) $-\frac{1}{3a}$

Ans. (4)

Sol. $w = \text{Area of parabola}$
 $= \frac{2}{3} (\text{Area of rectangle AC31A})$
 $= \frac{2}{3} P_0 (3-1) = \frac{4P_0}{3}$

When $V = 1$

$$(1-2)^2 = 4aP_0$$

$$P_0 = \frac{1}{4a}$$

$$w = \frac{4}{3} P_0 = \frac{4}{3} \frac{1}{4a} = \frac{1}{3a}$$

$$\boxed{w_{\text{gas}} = \frac{-1}{3a}}$$

38. For the two cells having same EMF E and internal resistance r , the current passing through the external resistor 6Ω is same when both the cells are connected either in parallel or in series. The value of internal resistance r is _____ Ω .

- (1) 3 (2) 4
 (3) 9 (4) 6

Ans. (4)

Sol. In series, $i_1 = \frac{2E}{6+2r}$

In parallel, $i_2 = \frac{E}{6+\frac{r}{2}}$

$$i_1 = i_2 \Rightarrow \frac{2E}{6+2r} = \frac{E}{6+\frac{r}{2}}$$

$$12+r = 6+2r$$

$$\boxed{r = 6\Omega}$$

39. Two wires A and B made of different materials of length 6.0 cm and 5.4 cm , respectively and area of cross sections $3.0 \times 10^{-5} \text{ m}^2$ and $4.5 \times 10^{-5} \text{ m}^2$, respectively are stretched by the same magnitude under a given load. The ratio of the Young's modulus of A to that of B is $x : 3$. The value of x is _____.

- (1) 1 (2) 4
 (3) 2 (4) 5

Ans. (4)

Sol. $T = \frac{F/A}{\Delta l/l} \Rightarrow Y = \frac{F l}{A \Delta l}$

$$\frac{Y_A}{Y_B} = \frac{l_A}{l_B} \left(\frac{A_B}{A_A} \right)$$

$$= \frac{6}{5.4} \left(\frac{4.5 \times 10^{-5}}{3 \times 10^{-5}} \right) = \frac{9}{5.4} = \frac{5}{3} \Rightarrow \frac{x}{3} = \frac{5}{3}$$

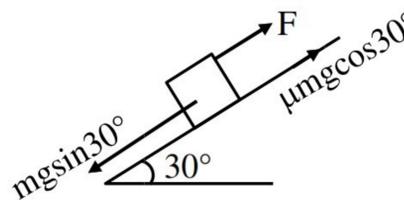
$$\boxed{x = 5}$$

40. A block of mass 5 kg is moving on an inclined plane which makes an angle of 30° with the horizontal. Friction coefficient between the block and inclined plane surface is $\frac{\sqrt{3}}{2}$. The force to be applied on the block so that the block will move down without acceleration is _____ N .

- (1) 25 (2) 12.5
 (3) 7.5 (4) 15

Ans. (2)

Sol.



$$mg \sin 30^\circ = F + \mu mg \cos 30^\circ$$

$$F = 5 \times 10 \times \frac{1}{2} - \frac{\sqrt{3}}{2} \times 5 \times 10 \times \frac{\sqrt{3}}{2}$$

$$F = 25 - \frac{75}{2} = 25 - 37.5$$

$$F = -12.5 \text{ N}$$

\therefore force will be downward on incline of magnitude 12.5 N

41. Given below are two statements :

Statement-I : A plane wave after passing through prism remains as plane wave but passing through small pin hole may become spherical wave.

Statement-II : The curvature of a spherical wave emerging from a slit will increase for increasing slit width.

In the light of the above statements, choose the **correct** answer from the options given below.

- (1) Both Statement-I and Statement-II are false.
 (2) Both Statement-I and Statement-II are true.
 (3) Statement-I is true but Statement-II is false.
 (4) Statement-I is false but Statement-II is true.

Ans. (3)

SECTION-B

46. The displacement of a particle, executing simple harmonic motion with time period T , is expressed as $x(t) = A \sin \omega t$, where A is the amplitude. The maximum value of potential energy of this oscillator is found at $t = T/2\beta$. The value of β is _____.

Ans. (2)

Sol. Potential energy is maximum at extreme position
The particle starting at mean position reaches extreme position in time $\frac{T}{4}$.

47. The ratio of de Broglie wavelength of a deuteron with kinetic energy E to that of an alpha particle with kinetic energy $2E$, is $n : 1$. The value of n is _____.

(Assume mass of proton = mass of neutron) :

Ans. (2)

Sol. $\lambda = \frac{h}{mv} = \frac{h}{\sqrt{2m \cdot KE}}$

$$\frac{\lambda_d}{\lambda_\alpha} = \sqrt{\frac{m_\alpha \cdot KE_\alpha}{m_d \cdot KE_d}} = \sqrt{\frac{4m \cdot 2E}{2m \cdot E}} = 2 : 1$$

48. A solid sphere of radius 10 cm is rotating about an axis which is at a distance 15 cm from its centre. The radius of gyration about this axis is \sqrt{n} cm. The value of n is

Ans. (265)

Sol. Let radius of gyration is k
 $\Rightarrow mk^2 = \frac{2}{3}mR^2 + md^2$

$$k^2 = \frac{2}{3} \times 10^2 + 15^2 = 265$$

$$(\sqrt{n})^2 = 265 \Rightarrow n = 265$$

49. A convex lens of refractive index 1.5 and focal length $f = 18$ cm is immersed in water. The difference in focal lengths of the given lens when it is in water and in air is $\alpha \times f$. The value of α is _____.

(refractive index of water = 4/3)

Ans. (3)

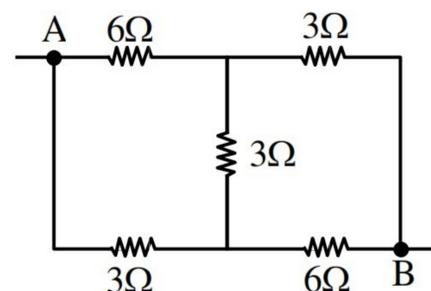
Sol. $\frac{1}{f_{\text{Air}}} = \left(\frac{1.5-1}{1}\right)\left(\frac{1}{R_1} + \frac{1}{R_2}\right)$

$$\frac{1}{f_{\text{water}}} = \left(\frac{1.5-4/3}{4/3}\right)\left(\frac{1}{R_1} + \frac{1}{R_2}\right)$$

$$\Rightarrow \frac{f_{\text{water}}}{f_{\text{air}}} = \frac{0.5}{0.5/4} = 4$$

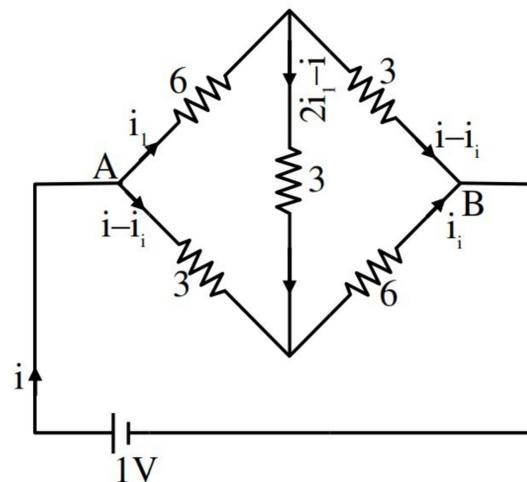
$$\Rightarrow f_{\text{water}} - f_{\text{air}} = 3f$$

50. The equivalent resistance between the points A and B in the following circuit is $\frac{x}{5} \Omega$. The value of x is _____.



Ans. (21)

Sol.



$$6i_1 + 3(2i_1 - i) = 3(i - i_1)$$

$$\Rightarrow 15i_1 = 6i \Rightarrow i_1 = \frac{2}{5}i \quad \text{--- (1)}$$

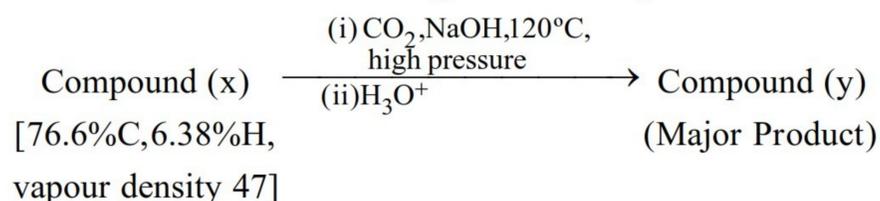
$$3(i - i_1) + 6i_1 = 1$$

$$3i + 3i_1 = 1$$

$$\left(3 + \frac{6}{5}\right)i = 1$$

$$\Rightarrow i = \frac{5}{21} \text{ A} = \frac{1\text{V}}{R_{\text{eq}}} \Rightarrow R_{\text{eq}} = \frac{21}{5} \Omega$$

54. Consider the following reaction sequence

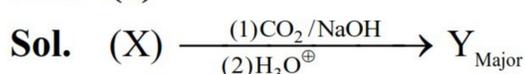


Compound (y) develops characteristic colour with neutral FeCl_3 solution.

Identify the **INCORRECT** statement from the following for the above sequence.

- (1) Both compounds x and y will dissolve in NaOH.
- (2) Compound y will dissolve in NaHCO_3 and evolve a gas.
- (3) Compound x is more acidic than compound y.
- (4) Both compounds x and y will burn with sooty flame.

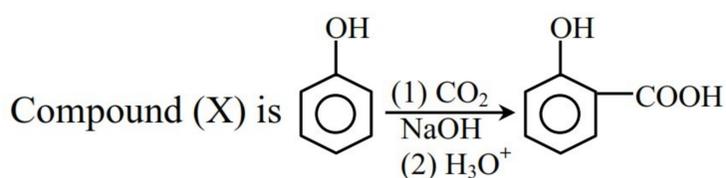
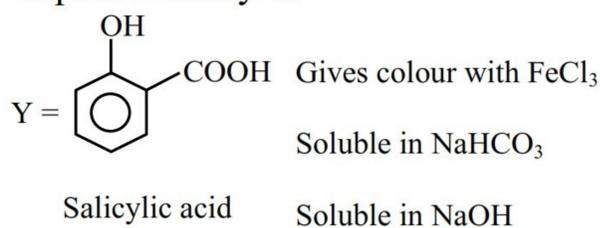
Ans. (3)



76.6% C

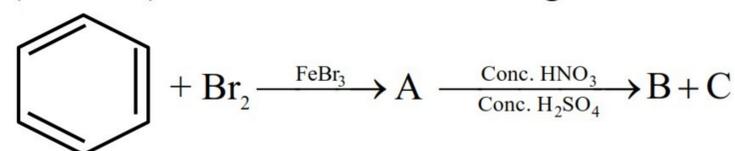
6.38% H

Vapour Density 47



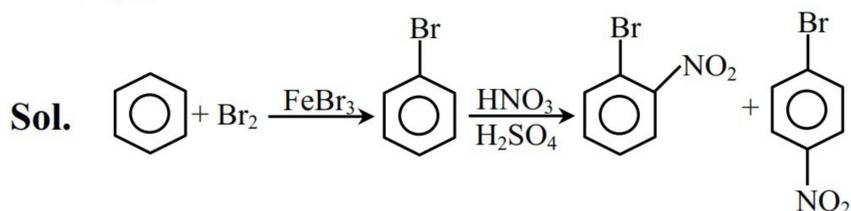
Kolbe Schmitt reaction

55. Method used for separation of mixture of products (B and C) obtained in the following reaction is:



- (1) Simple distillation
- (2) sublimation
- (3) steam distillation
- (4) fractional distillation

Ans. (4)



B & C separate by Fractional Distillation method Due to their different boiling point.

56. Consider a weak base 'B' of $\text{pK}_b = 5.699$. 'x' mL of 0.02 M HCl and 'y' mL of 0.02 M weak base 'B' are mixed to make 100 mL of a buffer of pH 9 at 25°C . The values of 'x' and 'y' respectively are:

[Given: $\log 2 = 0.3010$, $\log 3 = 0.4771$, $\log 5 = 0.699$]

(1)

x	y
11.1	88.9

(2)

x	y
42.7	57.3

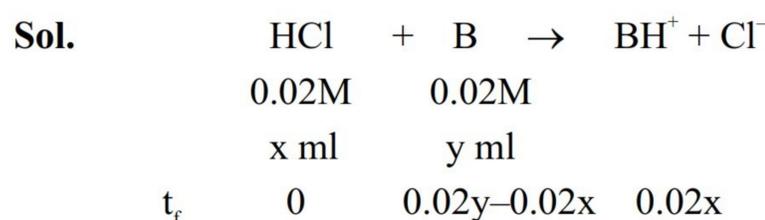
(3)

x	y
14.3	85.7

(4)

x	y
85.7	14.3

Ans. (3)



$$\text{pOH} = \text{pK}_b + \log \left[\frac{\text{Salt}}{\text{Base}} \right]$$

$$5 = 5.699 + \log \left[\frac{\text{Salt}}{\text{Base}} \right]$$

$$\frac{x}{y-x} = \frac{1}{5}$$

$$6x = y$$

$$7x = 100$$

$$x = \frac{100}{7} \text{ ml}$$

$$\& y = \frac{600}{7} \text{ ml}$$

57. An organic compound undergoes first order decomposition. The time taken for decomposition to $\left(\frac{1}{8}\right)^{\text{th}}$ and $\left(\frac{1}{10}\right)^{\text{th}}$ of its initial concentration are $t_{1/8}$ and $t_{1/10}$ respectively.

What is the value of $\frac{t_{1/8}}{t_{1/10}} \times 10$?

($\log 2 = 0.3$)

(1) 9

(2) 0.9

(3) 3

(4) 30

Ans. (1)

Sol. $t = \frac{1}{k} \ln \frac{A_0}{A_t}$

$$t_{1/8} = \frac{1}{k} \ln \frac{A_0}{A_0/8} = \frac{1}{k} \ln 8$$

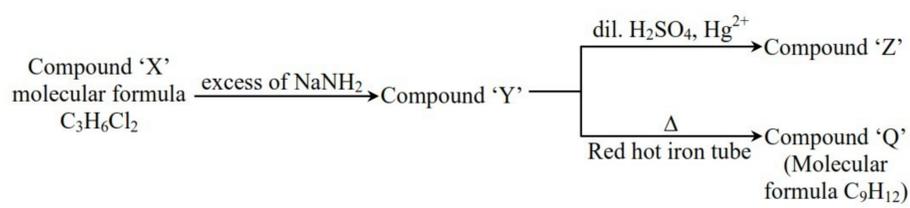
$$t_{1/10} = \frac{1}{k} \ln \frac{A_0}{A_0/10} = \frac{1}{k} \ln 10$$

$$\frac{t_{1/8}}{t_{1/10}} = \frac{\ln 8}{\ln 10} = \frac{\log 8}{\log 10}$$

$$\frac{t_{1/8}}{t_{1/10}} = \log 8 = 3 \log 2 = 0.9$$

$$\frac{t_{1/8}}{t_{1/10}} \times 10 = 9$$

58. Given below are two statements for the following reaction sequence.



Statement I: Compound 'Z' will give yellow precipitate with NaOI.

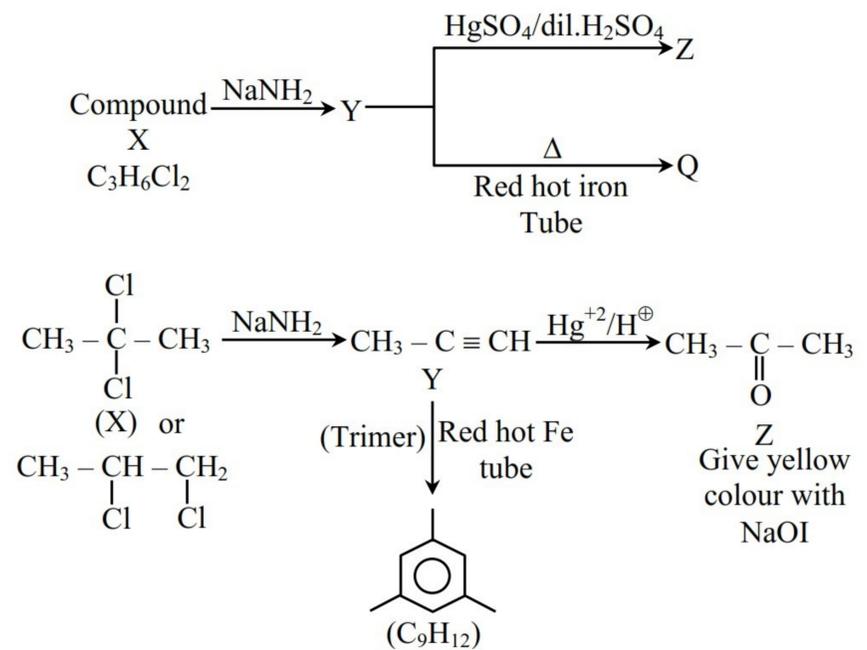
Statement II: Compound 'Q' has two different types of 'H' atoms (aromatic : aliphatic) in the ratio 1 : 3.

In the light of the above statements, choose the **correct** answer from the option given below:

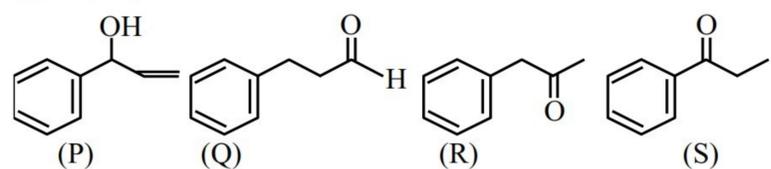
- (1) Statement I is true but Statement II is false
- (2) Both Statement I and Statement II are true
- (3) Statement I is false but Statement II is true
- (4) Both Statement I and Statement II are false

Ans. (2)

Sol.



59. Given below are the four isomeric compounds (P, Q, R, S)



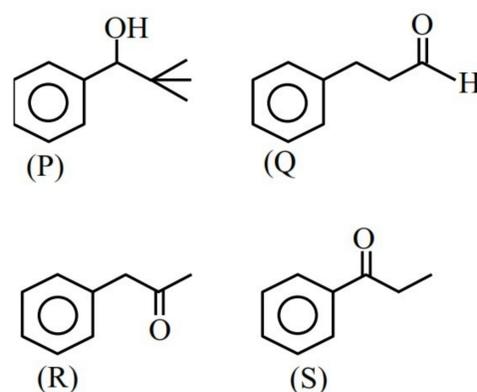
Identify **correct** statements from below.

- A. Q, R and S will give precipitate with 2, 4 - DNP.
 - B. P and Q will give positive Bayer's test.
 - C. Q and R will give sooty flame.
 - D. R and S will give yellow precipitate with $I_2/NaOH$.
 - E. Q alone will deposit silver with Tollen's reagent
- Choose the correct option.

- (1) A, C and E only
- (2) A and E only
- (3) C and E only
- (4) A, B, D and E only

Ans. (1)

Sol.



- (A) Q, R, S all three give 2, 4 DNP test as they have Aldehyde/ketone group
- (C) Q & R gives sooty flame
- (E) Q gives Tollens reagent test

60. Given below are two statements:

Statement I: The number of species among BF_4^- , SiF_4 , XeF_4 and SF_4 , that have unequal E-F bond lengths is two. Here, E is the central atom.

Statement II: Among O_2^- , O_2^{2-} , F_2 and O_2^+ , O_2^- has the highest bond order.

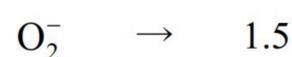
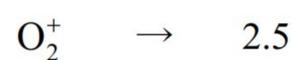
In the light of the above statements, choose the **correct** answer from the options given below

- (1) Both Statement I and Statement II are false
- (2) Both Statement I and Statement II are true
- (3) Statement I is true but Statement II is false
- (4) Statement I is false but Statement II is true

Ans. (1)

Sol. In BF_4^- , SiF_4 and XeF_4 all bond lengths are identical

Molecules B.O.



61. Regarding the hydrides of group 15 elements EH_3 (E = N, P, As, Sb), select the correct statement from the following:

- A. The stability of hydrides decreases down the group.
- B. The basicity of hydrides decreases down the group.
- C. The reducing character increases down the group.
- D. The boiling point increases down the group.

Choose the **correct** answer from the options given below:

(1) A, B & C only (2) A & D only

(3) A, B, C & D (4) B & C only

Ans. (1)

Sol. Stability : $\text{NH}_3 > \text{PH}_3 > \text{AsH}_3 > \text{SbH}_3 > \text{BiH}_3$

Basicity : $\text{NH}_3 > \text{PH}_3 > \text{AsH}_3 > \text{SbH}_3 > \text{BiH}_3$

Reducing character : $\text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3$

Boiling point : $\text{PH}_3 < \text{AsH}_3 < \text{NH}_3 < \text{SbH}_3 < \text{BiH}_3$

62. Given below are two statements:

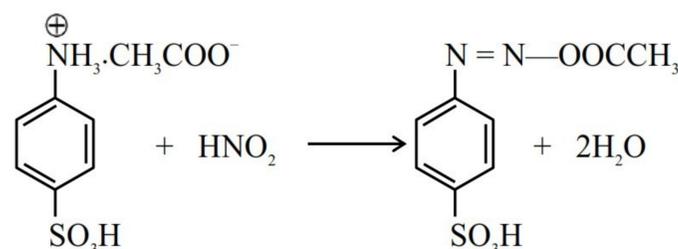
Statement I: Griss-Ilosvay test is used for the detection of nitrite ion, which involves the use of sulphanilic acid and α -naphthylamine reagent.

Statement II: In the above test, sulphanilic acid is diazotized by the acidified nitrite ion, which on further coupling with α -naphthylamine forms an azo-dye.

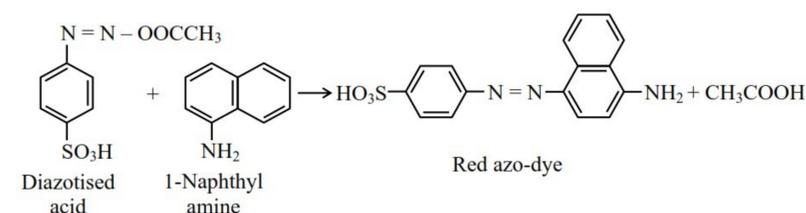
In the light of the above statements, choose the **correct** answer from the options given below

- (1) Statement I is false but Statement II is true
- (2) Both Statement I and Statement II are true
- (3) Statement I is true but Statement II is false
- (4) Both Statement I and Statement II are false

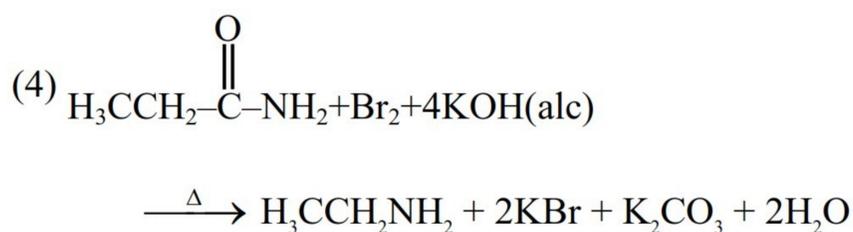
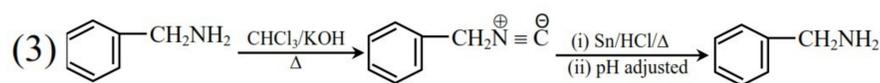
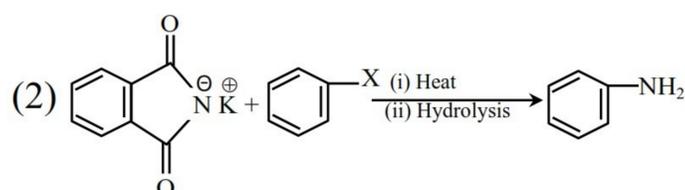
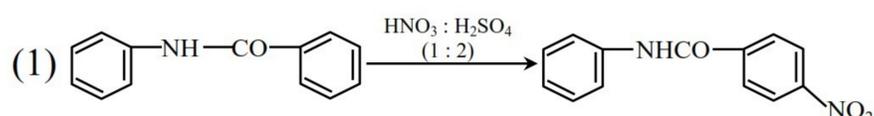
Ans. (2)



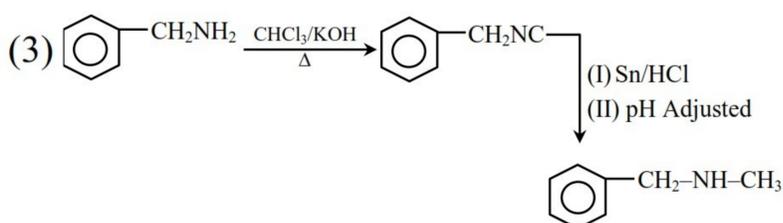
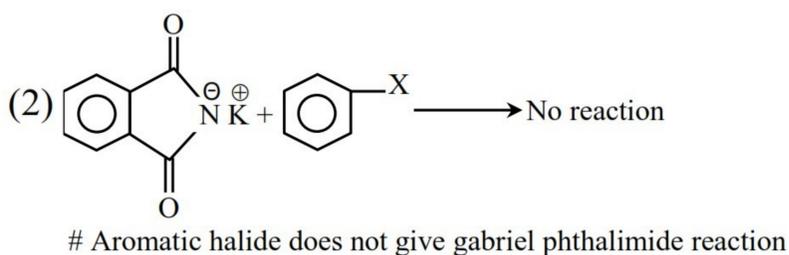
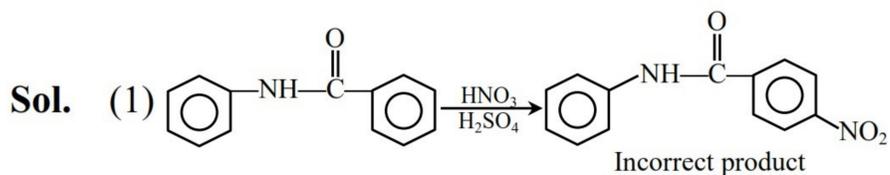
(Sulphanilic acid solution)



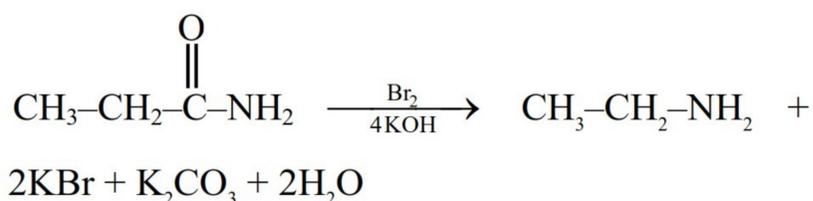
63. Consider the following reactions giving major product. Identify the correct reaction.



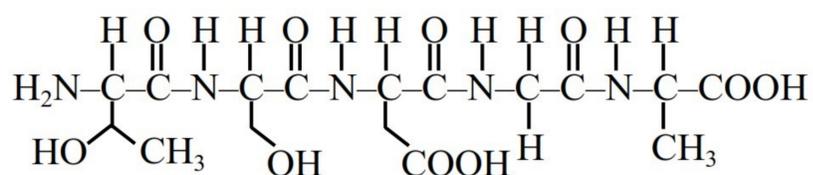
Ans. (4)



(4) Hoffmann bromamide degradation



64. In the given pentapeptide, find out an essential amino acid (Y) and the sequence present in the pentapeptide:



Choose the correct answer from the options given below:

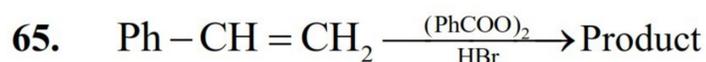
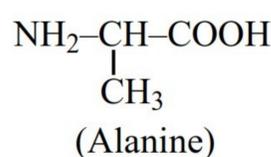
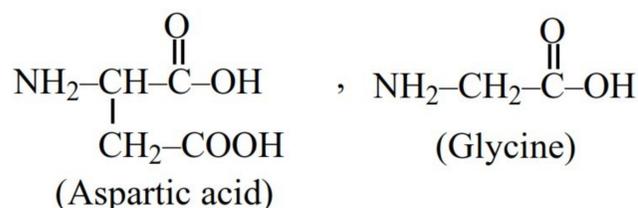
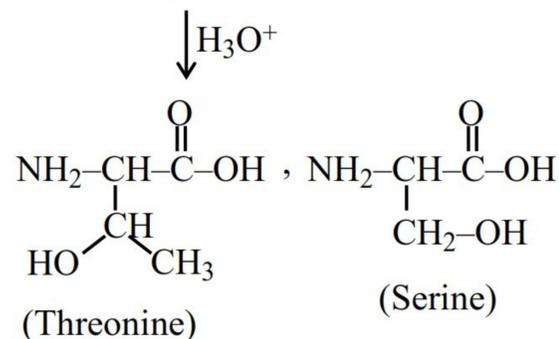
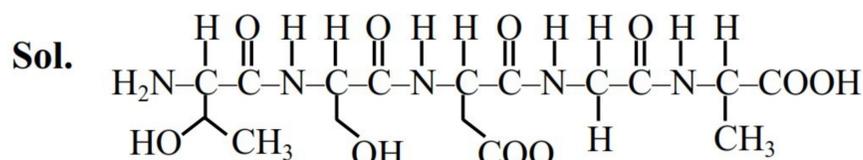
(1)	(Y)	(Sequence)
	Threonine	Ser - Thr - Asp - Gly - Ala

(2)	(Y)	(Sequence)
	Serine	Thr - Ser - Asp - Ala - Gly

(3)	(Y)	(Sequence)
	Threonine	Thr - Ser - Asp - Gly - Ala

(4)	(Y)	(Sequence)
	Serine	Ser - Asp - Thr - Ala - Gly

Ans. (3)



Consider the above reaction

- A. The reaction proceeds through a more stable radical intermediate.
 B. The role of peroxide is to generate $\dot{\text{H}}$ (Hydrogen radical).
 C. During this reaction, benzene is formed as a byproduct.
 D. 1-Bromo-2-phenylethane is formed as the minor product.

E. The same reaction in absence of peroxide proceeds via carbocation intermediate.

Identify the correct statements. Choose the *correct* answer from the options given below:

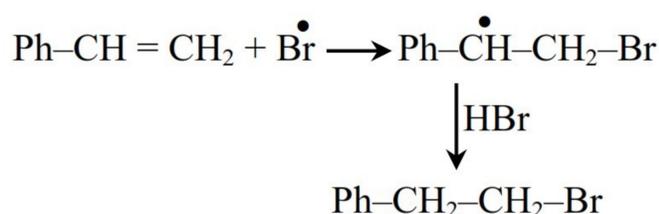
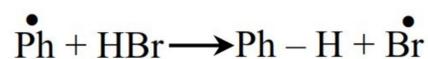
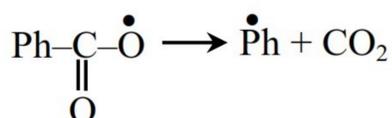
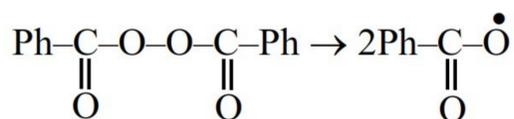
- (1) A & E Only
 (2) A, B & D Only
 (3) C, D & E Only
 (4) A, C & E Only

Ans. (4)



Anti Markovnikov addition

- Reaction follow radical addition in presence of peroxide
- In absence of peroxide follow carbocation mechanism
- Benzene also formed



66. The wave numbers of three spectral lines of H atom are considered. Identify the set of spectral lines belonging to Balmer series.

(R = Rydberg constant)

(1) $\frac{5R}{36}, \frac{3R}{16}, \frac{21R}{100}$

(2) $\frac{5R}{36}, \frac{8R}{9}, \frac{15R}{16}$

(3) $\frac{7R}{144}, \frac{3R}{16}, \frac{16R}{255}$

(4) $\frac{3R}{4}, \frac{3R}{16}, \frac{7R}{144}$

Ans. (1)

Sol. Balmer series line $\Rightarrow \bar{\nu} = R_H Z^2 \left[\frac{1}{2^2} - \frac{1}{n^2} \right]$

if $n = 3 \Rightarrow \bar{\nu} = R(1)^2 \left[\frac{1}{2^2} - \frac{1}{3^2} \right] = \frac{5R}{36}$

if $n = 4 \Rightarrow \bar{\nu} = \frac{3R}{16}$

if $n = 5 \Rightarrow \bar{\nu} = \frac{21R}{100}$

67.

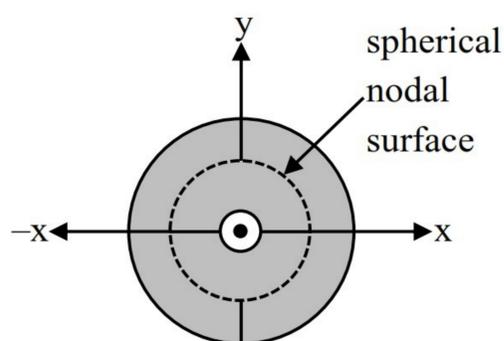


Figure 1. electron probability density for 2s orbital

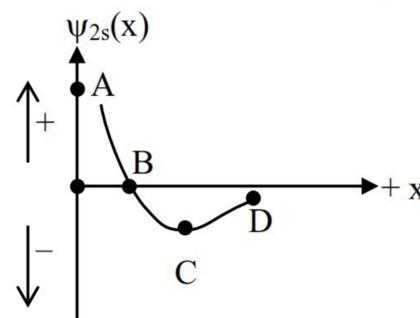


Figure 2. wave function for 2s orbital

Which of the following point in Figure 2 most accurately represents the nodal surface as shown in Figure 1 ?

- (1) B (2) D
(3) C (4) A

Ans. (1)

Sol. At spherical node

$$\psi_r = 0$$

68. Given below are two statements :

Statement I : The number of pairs, from the following, in which both the ions are coloured in aqueous solution is 3.



Statement II : Th^{4+} is the strongest reducing agent among $\text{Th}^{4+}, \text{Ce}^{4+}, \text{Gd}^{3+}$ and Eu^{2+} .

In the light of the above statements, choose the **correct** answer from the options given below

- (1) Statement I is true but Statement II is false
(2) Statement I is false but Statement II is true
(3) Both Statement I and Statement II are false
(4) Both Statement I and Statement II are true

Ans. (3)

Sol. $\text{Sc}^{3+}, \text{Ti}^{4+}$ and Zn^{2+} are colourless
 Th^{4+} cannot act as a reducing agent.

69. In period 4 of the periodic table, the elements with highest and lowest atomic radii are respectively.

- (1) Na & Cl
- (2) K & Se
- (3) K & Br
- (4) Rb & Br

Ans. (3)

Sol. In a period moving from left to right atomic size decreases.

70. The correct statement among the following is :

- (1) $[\text{Ni}(\text{CN})_4]^{2-}$ and $[\text{NiCl}_4]^{2-}$ are diamagnetic and $\text{Ni}(\text{CO})_4$ is paramagnetic.
- (2) $\text{Ni}(\text{CO})_4$ and $[\text{NiCl}_4]^{2-}$ are diamagnetic and $[\text{Ni}(\text{CN})_4]^{2-}$ is paramagnetic.
- (3) $\text{Ni}(\text{CO})_4$ and $[\text{Ni}(\text{CN})_4]^{2-}$ are diamagnetic and $[\text{NiCl}_4]^{2-}$ is paramagnetic.
- (4) $\text{Ni}(\text{CO})_4$ is diamagnetic and $[\text{NiCl}_4]^{2-}$ and $[\text{Ni}(\text{CN})_4]^{2-}$ are paramagnetic.

Ans. (3)

Sol. $[\text{Ni}(\text{CN})_4]^{2-} \rightarrow 3d^8 \rightarrow \text{diamagnetic} \rightarrow dsp^2$

$[\text{Ni}(\text{CO})_4] \rightarrow 3d^{10} \rightarrow \text{diamagnetic} \rightarrow sp^3$

$[\text{NiCl}_4]^{2-} \rightarrow 3d^8 \rightarrow e^{2,2} t_2^{2,1,1} \rightarrow sp^3 \rightarrow \text{paramagnetic}.$

SECTION-B

71. 500 mL of 1.2 M KI solution is mixed with 500 mL of 0.2 M KMnO_4 solution in basic medium. The liberated iodine was titrated with standard 0.1 M $\text{Na}_2\text{S}_2\text{O}_3$ solution in the presence of starch indicator till the blue color disappeared. The volume (in L) of $\text{Na}_2\text{S}_2\text{O}_3$ consumed is _____. (Nearest integer)

Ans. (3)

Sol. $\text{MnO}_4^- + \text{I}^- \rightarrow \text{MnO}_2 + \text{I}_2$

$\text{I}_2 + \text{S}_2\text{O}_3^{2-} \rightarrow \text{S}_4\text{O}_6^{2-} + \text{I}^-$

gram eq of $\text{KMnO}_4 = \text{gram eq of Na}_2\text{S}_2\text{O}_3$

$$0.2 \times \frac{500}{1000} \times 3 = 0.1 \times V \times 1$$

$$V = 3 \text{ L}$$

72. Consider the following redox reaction taking place in acidic medium

$\text{BH}_4^-(\text{aq}) + \text{ClO}_3^-(\text{aq}) \rightarrow \text{H}_2\text{BO}_3^-(\text{aq}) + \text{Cl}^-(\text{aq})$

If the Nernst equation for the above balanced reaction is

$$E_{\text{cell}} = E_{\text{cell}}^{\circ} - \frac{RT}{nF} \ln Q,$$

Then the value of n is _____. (Nearest integer)

Ans. (24)

Sol. $3\text{BH}_4^- + 4\text{ClO}_3^- \rightarrow 4\text{Cl}^- + 3\text{H}_2\text{BO}_3^- + 3\text{H}_2\text{O}$

n-factor = 8

moles = 3

$$\therefore n = 3 \times 8 = 24$$

73. X is the number of geometrical isomers exhibited by $[\text{Pt}(\text{NH}_3)(\text{H}_2\text{O})\text{BrCl}]$.

Y is the number of optically inactive isomer(s) exhibited by $[\text{CrCl}_2(\text{ox})_2]^{3-}$

Z is the number of geometrical isomers exhibited by $[\text{Co}(\text{NH}_3)_3(\text{NO}_2)_3]$

The value of X + Y + Z is _____.

Ans. (6)

Sol. Here

X = 3 (Two cis + one trans isomers)

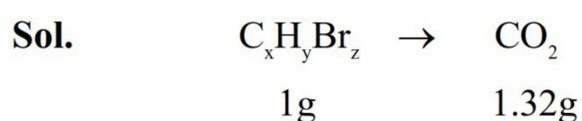
Y = 1 (trans isomer)

Z = 2 (Fac- mer isomer)

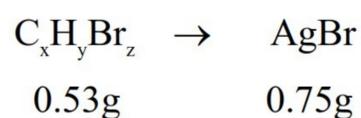
$$X+Y+Z = 3 + 1 + 2 = 6$$

74. 0.53 g of an organic compound (x) when heated with excess of nitric acid (concentrated) and then with silver nitrate gave 0.75 g of silver bromide precipitate. 1.0 g of (x) gave 1.32 g of CO₂ gas on combustion. The percentage of hydrogen in the compound (x) is _____. [Nearest Integer]
 [Given : Molar mass in g mol⁻¹ H : 1, C : 12, Br : 80, Ag : 108, O : 16; Compound (x) : C_xH_yBr_z]

Ans. (4)



$$\%C = \frac{1.32 \times 12}{44 \times 1} \times 100 = 36\%$$



$$\%Br = \frac{0.75 \times 80}{188 \times 0.53} \times 100 = 60.2\%$$

$$\%H = 100 - (36 + 60.2)$$

$$\%H \approx 4\%$$

75. Consider the dissociation equilibrium of the following weak acid $HA \rightleftharpoons H^+(aq) + A^-(aq)$

If the pK_a of the acid is 4, then the pH of 10 mM HA solution is _____. (Nearest integer)

[Given : The degree of dissociation can be neglected with respect to unity]

Ans. (3)

Sol. $pH = \frac{1}{2}[pK_a - \log c]$

$$pH = \frac{1}{2}[4 - \log 10^{-2}]$$

$$pH = 3$$